Biochemical Systems as Reactive Systems

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Processes and Functions







Many inputs and outputs



$P = ?in_1(x); ?in_2(y); !out_1(x+y); P$ $\oplus ?in_3(z); !out_2(\sqrt{z}); !out_1(2z); 0$



That's π -calculus

- To compose processes P we need:
 - Composition:
 - \circ Channel cration: (v x) P
 - **Recursion**:

*P

P | P

(with identity elem. 0) (with x bound in P) (equal to P | *P)

To execute actions we need:

• Channel reading: ?c(x); P (with x bound in P) • Channel writing: !c(M); P (with message M) • Choice: $P \oplus P$ (with identity elem. 0)

... and channels can be sent as messages!

Generalizing Functions and Automata

Unlike functions...

- Processes have multiple, explicitly named, input and output channels.
- Processes can run in *parallel*, can *deadlock* on their inputs, and can be *nondeterministic* in their outputs.

• Unlike automata (FSA)...

- Processes can transmit data (not just change state).
- While automata 'talk' to input strings, processes 'talk' to other processes: processes are communicating automata.
- Processes are not "finite state"; they can express unbounded computation in time (divergence) and space (proliferation).

Algebraic Properties

- Functions have one binder and one rule:
 Function application:
 - If $f(x) =_{def} M\{x\}$ then $f(a) = M\{a/x\}$
- Processes have two binders and two rules:

 Communication (input '?' binder)
 (?c(x);P{x}) ⊕ P' | (!c(a);Q) ⊕ Q' = P{a/x} | Q
 - Scope extrusion (new 'v' binder) If x not occurring in Q then ((v x)P) | Q = (v x)(P|Q)

Implementations

\square SPiM (Stochastic Pi Machine) •

- http://lepton.research.microsoft.com/VisualSPiM/ 0
- Runs in a browser with Silverlight. 0



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Processes and Chemistry

Continuous Chemical Systems

Reactions:

 Degradation Asymmetric Collision Symmetric Collision

Continuous reaction kinetics, respectively:

 $[A]^{\bullet} = -r[A]$ Exponential Decay $[A_i]^{\bullet} = -r[A_1][A_2]$ Mass Action Law $[A]^{\bullet} = -2r[A]^2$ Mass Action Law
(assuming A≠B_i≠A_j for all i,j)

π -calculus for Chemistry

To compose *soups* P we need:

- Stochastic channels: $(v x_r) P$
- Composition:
- P | P

• **Recursion**:

- *P
- r is the rate of an exponential distribution: the rate of communication on that channel
- (with identity elem. 0)
- (equal to P | *P)

To execute *species* we need:

?x_r; P

• Collision:

- (with no input variables)
- **Co-collision**:
- **Delay**:
- Choice:
- !x_r; P (with no output messages)
 - τ_r ; P (= (v x_r) ?x_r; P|!x_r; 0 for any x not in P)
- $P \oplus P$ (with identity elem. 0)

Discrete Chemical Systems (1)

Reaction:

 $A \rightarrow^{r} B_{1} + \ldots + B_{n}$

Discrete reaction kinetics:

 $A = \tau_r; (B_1 | ... | B_n)$

The mathematical meaning of that is a Continuous Time Markov Chain (for a specific set of initial conditions, e.g. a single A molecule), here represented as a transition graph:



Hence the π -calculus description abstracts from initial conditions (like ODEs). For each set of initial conditions, a CTMC can be systematically extracted from the stochastic π -calculus models.

Discrete Chemical Systems (2)

(Uniquely named) reaction:

c: $A_1 + A_2 \rightarrow^r B_1 + \dots + B_n$

Discrete reaction kinetics:

 $A_{1} = ?c_{r}; (B_{1}|...|B_{i})$ $A_{2} = !c_{r}; (B_{i}|...|B_{n})$

(the name of the reaction becomes the channel) (splitting results is arbitrary: $1 \le i \le n$)

With initial conditions $A_1 | A_2$ (single molecules), the CTMC is:



Discrete Chemical Systems (3)

(Uniquely named) reaction:

c: $A + A \rightarrow^r B_1 + \dots + B_n$

Discrete reaction kinetics:

 $A = ?c_{r/2}; (B_1 | ... | B_i) \oplus !c_{r/2}; (B_i | ... | B_n)$ $1 \le i \le n$

With initial conditions A|A (two molecules), the CTMC is as follows; note that each copy of A can do an input or an output, so there are two possible paths to the outcome:



From Reactions to Processes



That Chemical System in SPiM

 k_2

 \mathbf{k}_1



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$$F = ?v_{4(k4/2)}; B \oplus !v_{4(k4/2)}; 0$$



Gillespie-style stochastic simulation

directive sample 10.0
directive plot A(); B(); C(); D(); E(); F()

val k1 = 0.001 new v1@k1:chan val k2 = 0.001 new v2@k1:chan val k3 = 1.0

val k4 = 0.001 new v4@k4/2.0:chan

let A() = do ?v1;(C()|C()) or ?v2;D()
and B() = !v1
and C() = do !v2 or delay@k3;(E()|F())
and D() = ()

unu D() - ()

and E() = ()

and F() = do ?v4;B() or !v4

run 300 of (A()|B()|C()|D()|E()|F())

Modeling Techinques

- That is a *systematic* way to translate reactions to processes.
- But there can be *better* ways to do it.
- That is, ways that produce more compact and/or modular models, but with the same kinetics.

Ex: Catalysis

- Two reactions, same catalyst C
 - According to the general scheme the catalyst uses one channel for each reaction it catalyzes

a:
$$A + C \rightarrow^{r} C + B$$
 $C = !a_{r}; C \oplus !b_{r}; C$

b: $D + C \rightarrow^{r} C + E$ $A = ?a_{r}; B$

 $D = ?b_r; E$

 Modularizing: the catalyst has its own catalysis channel c, used for all the reactions it catalyzes:

Processes and Biochemistry

π -calculus for Biochemistry

- Biochemistry here means: direct modeling of complexation and polymerization.
- We now go back to the full (and stochastic) π– calculus: we need to pass channels as messages!

Complexation

$A + B \xrightarrow{s} A:B$

There is no good notation for this reaction in chemistry: A:B is considered as a separate species (which leads to combinatorial explosion of models).

But there is a way to write this precisely in π -calculus. There is a single public *association* channel a_r at rate r, and many private *dissociations* channels d_s at rate s, one for each complexation event (created by v):

$$\begin{array}{ll} \mathsf{A}_{\mathsf{free}} &= (v \ \mathsf{d}_{\mathsf{s}}) \ ! \mathsf{a}_{\mathsf{r}}(\mathsf{d}_{\mathsf{s}}); \ \mathsf{A}_{\mathsf{bound}}(\mathsf{d}_{\mathsf{s}}) \\ \mathsf{A}_{\mathsf{bound}}(\mathsf{d}_{\mathsf{s}}) &= ! \mathsf{d}_{\mathsf{s}}; \ \mathsf{A}_{\mathsf{free}} \end{array}$$

 $\begin{array}{ll} B_{free} & = ?a_r(d_s); \ B_{bound}(d_s) \\ B_{bound}(d_s) & = ?d_s; \ B_{free} \end{array}$

Note that we are describing A *independently* of B: as in the catalysis example, A could form complexes with many different species over the a_r channel.

More compactly:

Polymerization

Polymerization is iterated complexation

- It can be represente in π -calculus *finitely*, with one process (definition) for each monomer.
- Note that polymerization cannot be described *finitely* in chemistry (or ODEs) because there it needs one reaction for each *length* of polymer.
- The reason it works in π -calculus is because of the v operator. It enables the finite representation of systems of potentially unbounded complexity.
- Like real biochemistry, where the structure of each monomer is coded in a finite piece of DNA, and yet unbounded-length polymers happen.

Processes and Genes

Higher-level Modeling

• The modeling level is up to you

- We have seen how to use π -calculus to model chemical and biochemical events.
- But it can be used to model any kind of 'events':
 - A process can represent: a molecule, a gene, a cell, an organism, etc.
 - A communication on a channel can represent: a molecular collision, a transcripion activation, a communication between cells, an interaction between organism (epidemics), etc.
- Let's see how to model gene networks
 - No longer single-molecule interactions, but rather interactions between genes and transcription factors.

The Classical ODE Approach

[Chen, He, Church]



$$\frac{d\mathbf{r}}{dt} = f(\mathbf{p}) - V\mathbf{r}$$
$$\frac{d\mathbf{p}}{dt} = L\mathbf{r} - U\mathbf{r}$$

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n: number of genesr mRNA concentrations (n-dim vector)p protein concentrations (n-dim vector)

f(p) transcription functions:(n-dim vector polynomials on p)



A stochastic rate r is always associated with each channel a_r (at channel creation time) and delay t_r , but is often omitted when unambiguous.







Signal Amplification





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pos(a,b) | pos(b,c)

even with no a input, constitutive production of b gets amplified to a high c signal

Signal Normalization



a non-zero input level, a, whether weak or strong, is renormalized to a standard level, c.

neg(a,b) | neg(b,c)



^{30*}tr(a) | neg(a,b) | neg(b,c)







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Conclusions

Conclusions

π-calculus

- A mathematical notation for reactive systems
- In stochastic form, suitable for representing discrete chemistry, biochemistry, etc.
- Some unique properties: ability to finitely express systems of unbounded complexity, like networks of complexing proteins.

Further Reading

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