# Molecules as Automata

Representing Biochemical Systems as Collectives of Interacting Automata

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### Engineering Method



Direct Engineering (Synthetic Biology)

### Scientific Method



Reverse Engineering (Systems Biology)

### Engineering Method



Direct Engineering

### Scientific Method



**Reverse Engineering** 











# Modeling Approach

- We believe that {petri nets, process algebra, term rewriting, multiagent systems} are {better, complementary} for modeling biological systems than {SBML, Kohn charts, chemical reactions, ODEs}.
- We take a paper from the literature (usually ODEs or chemical reactions) and "code it up" in e.g. Petri nets.
- How do we know that's the "same system"? How do we convince mathematical biologists that we are doing the "right thing"?

# (Macro-) Molecules as (Interacting) Automata

# **Process Algebra**

- Reactive systems (living organisms, computer networks, operating systems, ...)
  - Math is based on *entities that react/interact with their environment* ("processes"), not on functions from domains to codomains.
- Concurrent
  - Events (reactions/interactions) happen concurrently and asynchronously, not sequentially like in function composition.
- Stochastic
  - Or probabilistic, or nondeterministic, but is never about deterministic system evolution.
- Stateful
  - Each concurrent activity ("process") maintains its own local state, as opposed to stateless functions from inputs to outputs.
- Discrete
  - Evolution through discrete transitions between discrete states, not incremental changes of continuous quantities.
- Kinetics of interaction
  - An "interaction" is anything that moves a system from one state to another.

# **Interacting Automata**



#### Kinetic laws:

# **Interacting Automata**



Kinetic laws:

Two complementary actions may result in an interaction.

# **Interacting Automata**



Kinetic laws:

Two complementary actions may result in an interaction.

A decay may happen spontaneously.

# **Interactions in a Population**



# **Interactions in a Population**



# **Interactions in a Population**



# Interactions in a Population (2)



# Interactions in a Population (2)



## **CTMC** Semantics



2r<sub>b</sub> {2A,1B}

CTMC

 $2r_{b}$ 



# **Reactions vs. Components**



## **Some Devices**



#### Ultrasensitive Switch



#### Cascade Amplifier



#### Symmetric Wave Generator



## **More Devices**

SPiM

A0

В0 С0

0.03





#### Repressilator (1 of 3 similar gates)





# **Semantics of Collective Behavior**

# The Two Semantic Sides of Chemistry



These diagrams commute via appropriate maps.

L. Cardelli: "On Process Rate Semantics" (TCS)

L. Cardelli: "A Process Algebra Master Equation" (QEST'07)

# **Quantitative Process Semantics**



# From CGF to Chemistry

# **Chemical Reactions**

Homeo Reaction

A  $\rightarrow^r B_1 + \dots + B_n (n \ge 0)$  $A_1 + A_2 \rightarrow^r B_1 + \dots + B_n$  (n  $\ge 0$ ) Hetero Reaction  $A + A \longrightarrow^r B_1 + \dots + B_n \quad (n \ge 0)$ 

Unary Reaction

d[A]/dt = -r[A]

 $d[A_{i}]/dt = -r[A_{1}][A_{2}]$ 

 $d[A]/dt = -2r[A]^2$ 

**Exponential Decay** 

Mass Action Law

Mass Action Law

(assuming  $A \neq B_i \neq A_i$  for all i,j)

#### No other reactions!

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The chemical Langevin equation Daniel T. Gillespie<sup>a)</sup>

Research Department, Code 4T4100D, Naval Air Warfare Center, China Lake, California 93555

Genuinely trimolecular reactions do not physically occur in dilute fluids with any appreciable frequency. Apparently trimolecular reactions in a fluid are usually the combined result of two bimolecular reactions and one monomolecular reaction, and involve an additional short-lived species.

#### Trimolecular reactions:

 $A + B + C \rightarrow^{r} D$ 

the measured "r" is an (imperfect) aggregate of e.g.:

 $A + B \leftrightarrow AB$ 

 $AB + C \rightarrow D$ 

Chapter IV: Chemical Kinetics [David A. Reckhow , CEE 572 Course] reactions may be either elementary or non- elementary. <u>Elementary reactions</u> are those reactions that occur exactly as they are written, without any intermediate steps. These reactions almost always involve just one or two reactants <u>Non-elementary reactions</u> involve a series of two or more elementary reactions. Many complex environmental reactions are non- elementary. In general, reactions with an overall reaction order greater than two, or reactions with some non-integer reaction order are non-elementary.		THE COLLISION THEORY OF REACTION RATES www.chemguide.co.uk The chances of all this happening if your reaction needed a collision involving more than 2 particles are remote. All three (or more) particles would have to arrive at exactly the same point in space at the same time, with everything lined up exactly right, and having enough energy to react. That's not likely to happen very often!	
Enzymatic S _⊑→r F	reactions:		

the "r" is given by Michaelis-Menten (approximated steady-state) laws:

 $E + S \leftrightarrow ES$  $ES \rightarrow P + E$ 

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# Chemical Ground Form (CGF)



!b

2A and 2B



 $B = ?a_{(r)}; A \oplus \tau_{(s)}; A$ 



**B** = ?a;A  $\oplus \tau_{(s)}$ ;A



 $A = !a;A \oplus ?a;B$  $B = ?a;A \oplus \tau_{(s)};A$ 



Interacting Automata	Discrete Chemistry
initial states A   A     A	initial quantities #A <sub>0</sub>
A @r A	A → <sup>r</sup> A'
A ?a A' B !a B'	A+B → <sup>r</sup> A'+B'
?a A !a A' @r A"	A+A → <sup>2r</sup> A'+A"



# From CGF to Chemistry: Ch(E)

E ::= 0 : X=M, E	Reagents
$M ::= 0 : \pi; P \oplus M$	Molecules
P::=0 : X   P	Solutions
$\pi ::= \tau_{(r)} : ?a_{(r)} : !a_{(r)}$	Interactions (delay, input, output)
CGF ::= E,P	Reagents plus Initial Conditions

E.X.i ≝ the i-th Å-summand of the molecule M associated with the X reagent of E

Chemical reactions for E,P:

(N.B.: <...> are reaction tags to obtain multiplicity of reactions, and P is P with all the | changed to +)

Ch(E) := {(<X.i>: X →<sup>r</sup> P) s.t. E.X.i =  $\tau_{(r)}$ ;P} ∪ {(<X.i,Y.j>: X + Y →<sup>r</sup> P + Q) s.t. X≠Y, E.X.i = ?a<sub>(r)</sub>;P, E.Y.j = !a<sub>(r)</sub>;Q} ∪ {(<X.i,X.j>: X + X →<sup>2r</sup> P + Q) s.t. E.X.i = ?a<sub>(r)</sub>;P, E.X.j = !a<sub>(r)</sub>;Q}

Initial conditions for P:

Ch(P) := P

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# From Chemistry to CGF

# From Chemistry to CGF (by example)

x:  $B \rightarrow^{s} A$ b:  $A+B \rightarrow^{r} A+A$ c:  $A+A \rightarrow^{2r} A+B$ Unique reaction names


### From Chemistry to CGF (by example)

- x:  $B \rightarrow {}^{s} A$
- b:  $A+B \rightarrow^{r} A+A$
- c:  $A+A \rightarrow^{2r} A+B$



- 1: Fill the matrix by columns:
  - Degradation reaction  $v_i: X \rightarrow k_i P_i$ add  $\tau; P_i$  to  $\langle X, v_{ij} \rangle$ .

- x:  $B \rightarrow {}^{s} A$
- b:  $A+B \rightarrow^r A+A$
- c:  $A+A \rightarrow^{2r} A+B$

	X <sub>(s)</sub>	b <sub>(r)</sub>	C <sub>(r)</sub>
А		?;A A	
В	τ;Α	!;0	

#### 1: Fill the matrix by columns:

Degradation reaction  $v_i: X \rightarrow k_i P_i$ add  $\tau; P_i$  to  $\langle X, v_{ii} \rangle$ . Hetero reaction  $v_i: X+Y \rightarrow k_i P_i$ add ?;  $P_i$  to  $\langle X, v_i \rangle$  and !; 0 to  $\langle Y, v_i \rangle$ 

- x:  $B \rightarrow {}^{s} A$
- b:  $A+B \rightarrow^r A+A$
- c:  $A+A \rightarrow^{2r} A+B$

	x <sub>(s)</sub>	b <sub>(r)</sub>	C <sub>(r)</sub>
А		?;A A	?;A B !;0
В	τ;Α	!;0	

#### 1: Fill the matrix by columns:

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- x:  $B \rightarrow {}^{s} A$
- b:  $A+B \rightarrow^{r} A+A$
- c:  $A+A \rightarrow^{2r} A+B$

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2: Read the result by rows:

$$A = ?b_{(r)}; (A | A) \oplus ?c_{(r)}; (A | B) \oplus !c_{(r)}; 0$$
$$B = \tau_{(s)}; A \oplus !b_{(r)}; 0$$

	X <sub>(s)</sub>	b <sub>(r)</sub>	C <sub>(r)</sub>
А		?;A A	?;A B !;0
В	τ;A	!;0	



Α

В

- x:  $B \rightarrow {}^{s} A$
- b:  $A+B \rightarrow^{r} A+A$
- c:  $A+A \rightarrow^{2r} A+B$

#### 1: Fill the matrix by columns:

Degradation reaction  $v_i: X \rightarrow k_i P_i$ add  $\tau; P_i$  to  $\langle X, v_{ij} \rangle$ . Hetero reaction  $v_i: X+Y \rightarrow k_i P_i$ add ?;  $P_i$  to  $\langle X, v_i \rangle$  and !; 0 to  $\langle Y, v_i \rangle$ Homeo reaction  $v_i: X+X \rightarrow k_i P_i$ add ?;  $P_i$  and !; 0 to  $\langle X, v_i \rangle$ 

2: Read the result by rows:

$$A = ?b_{(r)}; A \oplus ?c_{(r)}; (A | B) \oplus !c_{(r)}; 0$$
$$B = \tau_{(s)}; A \oplus !b_{(r)}; A$$



- x:  $B \rightarrow {}^{s} A$
- b:  $A+B \rightarrow^{r} A+A$
- c:  $A+A \rightarrow^{2r} A+B$

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2: Read the result by rows:

$$A = ?b_{(r)}; A \oplus ?c_{(r)}; B \oplus !c_{(r)}; A$$
$$B = \tau_{(s)}; A \oplus !b_{(r)}; A$$





## From Chemistry to CGF: Pi(C)

v:  $X \rightarrow^r Y_1 + ... + Y_n + 0$ Unary Reactionv:  $X_1 + X_2 \rightarrow^r Y_1 + ... + Y_n + 0$ Binary Reaction

From uniquely-labeled (v:) chemical reactions C to a CGF Pi(C):

$$\begin{array}{lll} \mathsf{Pi}(\mathsf{C}) &= & \{(\mathsf{X} = \ \oplus ((\mathsf{v}: \mathsf{X} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C}) \ of \ (\tau_{(\mathsf{k})}; \mathsf{P}) & \oplus \\ & \oplus ((\mathsf{v}: \mathsf{X} + \mathsf{Y} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C} \ \text{and} \ \mathsf{Y} \neq \mathsf{X}) \ of \ (?\mathsf{v}_{(\mathsf{k})}; \mathsf{P}) & \oplus \\ & \oplus ((\mathsf{v}: \mathsf{Y} + \mathsf{X} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C} \ \text{and} \ \mathsf{Y} \neq \mathsf{X}) \ of \ (!\mathsf{v}_{(\mathsf{k})}; \mathsf{O}) & \oplus \\ & \oplus ((\mathsf{v}: \mathsf{X} + \mathsf{X} \to^{\mathsf{k}} \mathsf{P}) \in \mathsf{C}) \ of \ (?\mathsf{v}_{(\mathsf{k}/2)}; \mathsf{P} \oplus !\mathsf{v}_{(\mathsf{k}/2)}; \mathsf{O}) & ) \\ & & \texttt{s.t.} \ \mathsf{X} \ \text{is a species in C} \end{array}$$



# Discrete-State Semantics



## **Discrete State Equivalence**

- Def: 🗯 is equivalent CTMC's (isomorphic graphs with same rates).
- Thm: E 🗯 Ch(E)
- Thm: C 🗯 Pi(C)



- For each E there is an E' 22 E that is detangled (E' = Pi(Ch(E)))

#### **Interacting Automata = Discrete Chemistry**

This is enough to establish that the process algebra is really faithful to the chemistry.

But CTMC are not the "ultimate semantics" because there are still questions of when two different CTMCs are actually equivalent (e.g. "lumping").

The "ultimate semantics" of chemistry is the *Chemical Master Equation* (derivable from the Chapman-Kolmogorov equation of the CTMC).



# From Discrete to Continuous Chemistry

### **The Gillespie Conversion**

Discrete Chemistry	Continuous Chemistry	$\gamma = N_A V$	:M <sup>-1</sup>
initial quantities $#A_0$	initial concentration [A] <sub>0</sub>	ns with [A] <sub>0</sub> =#	Α <sub>0</sub> /γ
A, A'	$A \to^k A'$	with <mark>k = r</mark>	:S <sup>-1</sup>
A+B ⊶•r A'+B'	$A + B \rightarrow^k A' + B'$	with <mark>k = rγ</mark>	:M <sup>-1</sup> s <sup>-1</sup>
A+A ⊶r A'+A″	$A+A \rightarrow^k A'+A''$	with <mark>k = rγ/</mark> 2	:M <sup>-1</sup> s <sup>-1</sup>

V = interaction volume  $N_A =$  Avogadro's number

Think  $\gamma = 1$ i.e. V = 1/N<sub>A</sub>

M = mol·L<sup>-1</sup> molarity (concentration)



## $Cont_{\gamma}$ and $Disc_{\gamma}$

#### 4.2-3 Definition: Cont<sub>y</sub> and Disc<sub>y</sub>

For a volumetric factor  $\gamma:M^{-1}$ , we define a translation  $Cont_{\gamma}$  from a discrete chemical systems (C,P), with species X and initial molecule count  $\#X_0 = \#X(P)$ , to a continuous chemical systems (C,V) with initial concentration  $[X]_0 = V_X$ . The translation  $Disc_{\gamma}$  is its inverse, up to a rounding error  $\lceil \gamma[X]_0 \rceil$  in converting concentrations to molecule counts. Since  $\gamma$  is a global conversion constant, we later usually omit it as a subscript.

$Cont_{\gamma}(X \rightarrow^{r} P)$	$= X \rightarrow^{k} P$	with $k = r$ ,	r:s <sup>-1</sup>	k:s <sup>-1</sup>
$Cont_{\gamma}(X+Y \rightarrow^{r} P)$	$= X+Y \rightarrow^{k} P$	with $\mathbf{k} = \mathbf{r} \boldsymbol{\gamma}$	r:s <sup>-1</sup>	k:M <sup>-1</sup> s <sup>-1</sup>
$Cont_{\gamma}(X+X \rightarrow^{r} P)$	$= X + X \rightarrow^{k} P$	with $k = r\gamma/2$	r:s <sup>-1</sup>	k:M <sup>-1</sup> s <sup>-1</sup>
$Cont_{\gamma}(\#X_0)$	= [X] <sub>0</sub>	with $[X]_0 = #X_0/\gamma$	$X_0:mol$	[X] <sub>0</sub> :M
$Disc_{\gamma}(X \rightarrow^{k} P)$	$= X \rightarrow^{r} P$	with $r = k$ ,	k:s <sup>-1</sup>	r:s <sup>-1</sup>
$Disc_{\gamma}(X \to^{k} P)$ $Disc_{\gamma}(X+Y \to^{k} P)$	$= X \rightarrow^{r} P$ $= X+Y \rightarrow^{r} P$	with $r = k$ , with $r = k/\gamma$	k:s <sup>-1</sup> k:M <sup>-1</sup> s <sup>-1</sup>	1.0
		-		r:s <sup>-1</sup>

 $Ch_{\gamma} := Cont_{\gamma} \circ Ch$ 



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## Continuous-State Semantics (summary)



### **Continuous State Equivalence**

• Def:  $\approx$  is equivalence of polynomials over the field of reals.



- For each E there is an E'  $\approx$  E that is detangled (E' = Pi(Ch(E)))
- For each E in automata form there is an an E' ≈ E that is detangled and in automata form (E' = Detangle(E)).

## GMA ≠ CME



#### $A+A \rightarrow^{2r} A =? A+A \rightarrow^{r} 0$



(For conservation of mass, consider instead  $A+A \rightarrow^{2r} A+B$  vs.  $A+A \rightarrow^{r} B+B$ )

## **Continuous vs. Discrete Groupies**



directive sample 5.0 1000	directive sample 5.0 1000	directive sample 5.0 1000
directive plot B(); A()	directive plot B(); A()	directive plot B(); A()
new a@1.0:chan()	new a⊛1.0:chan()	new a⊛1.0:chan()
new b@1.0:chan()	new b⊛1.0:chan()	new b⊛1.0:chan()
let A() = do !a; A() or ?b; B()	let A() = do !a; A() or ?b; ?b; B()	let A() = do !a; A() or ?b; ?b; ?b; B()
and B() = do !b; B() or ?a; A()	and B() = do !b; B() or ?a; ?a; A()	and B() = do !b; B() or ?a; ?a; ?a; A()
let Ad() = !a; Ad()	let Ad() = !a; Ad()	let Ad() = !a; Ad()
and Bd() = !b; Bd()	and Bd() = !b; Bd()	and Bd() = !b; Bd()
run 2000 of A()	run 2000 of A()	run 2000 of A()
run 1 of (Ad()   Bd())	run 1 of (Ad()   Bd())	run 1 of (Ad()   Bd())

#### **Scientific Predictions**





After a while, all 4 states are almost equally occupied.

The 4 states are almost never equally occupied.

# **Chemistry and Beyond**

### Process Algebra is 'Bigger' than Chemistry



### Process Algebra is 'Bigger' than Chemistry



### Process Algebra is 'Bigger' than Chemistry



# On the Computational Power of Biochemistry

## joint work with Gianluigi Zavattaro

University of Bologna

in: Algebraic Biology '08

### **Biochemistry = Collision + Complexation**



• Complexation is what proteins "do", in contrast to simpler chemicals.



• Leading to a process algebra that we call the Biochemical Ground Form (BGF).

## What's the Difference?

Consider linear polymerization:



The "chemical program" for polymerization:

 $P_0 + M \rightarrow P_1$   $P_1 + M \rightarrow P_2$   $P_2 + M \rightarrow P_3$   $P_3 + M \rightarrow P_4$ 

• an infinite (non-)program

- an infinite set of species
- an infinite set of ODEs

 $P_{10757} + M \rightarrow P_{10758}$ Such specificity is unreal. But "nature's program" for polymerization has to fit e.g. in the genome, so it cannot be infinite! Clearly, nature must be using a different "language" than basic chemistry:

$$+$$
  $\rightarrow$   $\rightarrow$ 

molecule with convex patch + molecule with concave patch  $\rightarrow$ molecule with convex patch

- a finite program
- a local rule

## **Expressiveness of Biochemistry**

- Basic chemistry (FSRN, or CGF) is not Turing-complete
   By reduction to Petri Net reachability [Soleveichik&al.].
- Biochemistry (FSRN + complexation, or BGF) is Turing-complete.
  - $\circ~$  By an encoding of Random Access Machines, using polymers for registers.
- A relatively simple extension of our CGF automata
   But it is not as easy to find a corresponding extension of chemistry!
- More powerful process algebras of course *are* Turing complete
  - They (e.g.  $\pi$ -calculus) include BGF, but they also have mechanisms that are not directly biologically justifiable.
  - In BGF we have in a sense the minimal biologically-inspired extension of FSRN, and it is already Turing-complete.
- Intrinsic to biochemistry (but not to simple chemistry) is a Turingcomplete mechanism.

## Conclusions

### **Conclusions**

#### Process Algebra

- An extension of automata theory to populations of interacting automata
- Modeling the behavior of individuals in an arbitrary environment
- Compositionality (combining models by juxtaposition)
- Connections between modeling approaches
  - Connecting the discrete/concurrent/stochastic/molecular approach
  - o to the continuous/sequential/deterministic/population approach

#### Connecting syntax with semantics

- Syntax = model presentation (equations/programs/diagrams/blobs etc.)
- Semantics = state space (generated by the syntax)
- Ultimately, connections between analysis techniques
  - We need (and sometimes have) good semantic techniques to analyze state spaces (e.g. calculus, but also increasingly modelchecking)
  - But we need equally good syntactic techniques to structure complex models (e.g. compositionality) and analyze them (e.g. process algebra)
- A bright future for Computer Science and Logic in modern Biology
  - Biology needs good analysis techniques for discrete systems analysis
     (modal logics, modelchecking, causality analysis, abstract interpretation, ...)

