# Artificial Biochemistry

# Luca Cardelli

## **Microsoft Research**

Algorithmic Bioprocesses Leiden, 2007-12-04

http://LucaCardelli.name

## **Cells Compute**

- No survival without computation!
  - Finding food
  - Avoiding predators
- How do they compute?
  - Unusual computational paradigms.
  - Proteins: do they work like electronic circuits? or process algebra?
  - Genes: what kind of software is that?
- Signaling networks
  - Clearly "information processing"
  - They are "just chemistry": molecule interactions
  - But what are their principles and algorithms?
- Complex, higher-order interactions
  - MAPKKK = MAP Kinase Kinase Kinase: that which operates on that which operates on that which operates on protein.



Natl. Acad. Sci. USA, 93, 10078-10083.

# The View from Systems Biology

#### Epidermal Growth Factor Receptor Pathway Map

Karvas Oda (12), Yukiko Matsuoka (2), Hitsaki Kitano (12)( 3) Ia kura kua mila, Misatan kurana kura a kurani da kurani.



# **Stochastic Collectives**

# **Stochastic Collectives**

### • "Collective":

- A large set of interacting finite state automata:
  - Not quite language automata ("large set")
  - Not quite cellular automata ("interacting" but not on a grid)
  - Not quite process algebra ("collective behavior")
  - Cf. multi-agent systems and swarm intelligence

### • "Stochastic":

- Interactions have *rates* 
  - Not quite discrete (hundreds or thousands of components)
  - Not quite continuous (non-trivial stochastic effects)
  - Not quite hybrid (no "switching" between regimes)
- Very much like biochemistry
  - Which is a large set of stochastically interacting molecules/proteins
  - Are proteins finite state and subject to automata-like transitions?
    - Let's say they are, at least because:
    - Much of the knowledge being accumulated in Systems Biology is described as state transition diagrams [Kitano].



## Interacting Automata



Communicating automata: a graphical FSA-like notation for "finite state restriction-free  $\pi$ -calculus processes". Interacting automata do not even exchange values on communication.

The stochastic version has *rates* on communications, and delays.



"Finite state" means: no composition or restriction inside recursion. Analyzable by standard Markovian techniques, by first computing the "product automaton" to obtain the underlying finite Markov transition system. [Buchholz] *Interactions* have rates. Actions DO NOT have rates.

## Interactions in a Population



## Interactions in a Population



## Interactions in a Population



## Interactions in a Population (2)



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## Interactions in a Population (2)



## **CTMC** Semantics



2r<sub>b</sub> {2A,1B}

CTMC

 $2r_{b}$ 

## Chemistry vs. Automata



A Petri-Net-like representation. Precise and dynamic A compositional graphical representation (precise, but not modular, scalable, or maintainable. dynamic *and* modular) and the corresponding <u>calculus</u>.

# Emergent Collective Behavior

# **Groupies and Celebrities**





### A stochastic collective of celebrities:



Stable because as soon as a A finds itself in the majority, it is more likely to find somebody in the same state, and hence change, so the majority is weakened.



### A stochastic collective of groupies:



Unstable because within an A majority, an A has difficulty finding a B to emulate, but the few B's have plenty of A's to emulate, so the majority may switch to B. Leads to deadlock when everybody is in the same state and there is nobody different to emulate.

# Both Together

A way to break the deadlocks: Groupies with just a few Celebrities



### Regularity can arise not far from chaos

## Hysteric Groupies

We can get more regular behavior from groupies if they "need more convincing", or "hysteresis" (history-dependence), to switch states.



# Semantics of Collective Behavior

## The Two Semantic Sides of Chemistry



These diagrams commute (for the "Chemical Ground Form" process algebra). L. Cardelli: "On Process Rate Semantics" (TCS) L. Cardelli: "A Process Algebra Master Equation" (QEST'07)

## Quantitative Process Semantics



# Stochastic Processes & Discrete Chemistry



## **Chemical Reactions**

Elementary Reactions:		Reaction kinetics: [A] = concentration of A			
$A \longrightarrow^{r} B_1 + \dots + B_n  (n \ge 0)$	Unary Reaction	d[A]/dt = -r[A]	Exponential Decay		
$A_1 + A_2 \rightarrow^r B_1 + + B_n$ (n20)	Hetero Reaction	d[A <sub>i</sub> ]/dt = -r[A <sub>1</sub> ][A <sub>2</sub> ]	Mass Action Law		
$A + A \rightarrow^{r} B_1 + + B_n (n \ge 0)$	Homeo Reaction	d[A]/dt = -2r[A] <sup>2</sup>	Mass Action Law		
		(assuming A≠Bi≠Aj :	for all i,j)		

#### No other reactions!

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#### The chemical Langevin equation

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Genuinely *trimolecular* reactions do not physically occur in dilute fluids with any appreciable frequency. *Apparently* trimolecular reactions in a fluid are usually the combined result of two bimolecular reactions and one monomolecular reaction, and involve an additional short-lived species.

Trimolecular reactions:  $A + B + C \rightarrow^{r} D$ the measured "r" is an (imperfect) aggregate of e.g.:  $A + B \leftrightarrow AB$  $AB + C \rightarrow D$ 

#### Chapter IV: Chemical Kinetics [David A. Reckhow , CEE 572 Course]

... reactions may be either elementary or nonelementary. <u>Elementary reactions</u> are those reactions that occur exactly as they are written, without any intermediate steps. These reactions almost always involve just one or two reactants. ... <u>Non-elementary reactions</u> involve a series of two or more elementary reactions. Many complex environmental reactions are non-elementary. In general, reactions with an overall reaction order greater than two, or reactions with some non-integer reaction order are non-elementary.

 $S \xrightarrow{E} P$ 

Enzymatic reactions:

#### THE COLLISION THEORY OF REACTION RATES

www.chemguide.co.uk

The chances of all this happening if your reaction needed a collision involving more than 2 particles are remote. All three (or more) particles would have to arrive at exactly the same point in space at the same time, with everything lined up exactly right, and having enough energy to react. That's not likely to happen very often!

> *Reactions* have rates. Molecules *do not* have rates.

the "r" is given by Michaelis-Menten (approximated steady-state) laws:  $E + S \leftrightarrow ES$  $ES \rightarrow P + E$ 

## Chemical Ground Form (CGF)





**?**a

## Automata to Chemistry

			V = interaction volume
Automata 🗕	<ul> <li>Discrete</li> <li>Chemistry - (molecule counts)</li> </ul>	$\begin{array}{l} \textbf{Continuous} \\ \hline \textbf{Chemistry} \\ (concentrations) \end{array} \gamma = N_A V$	$N_A$ = Avogadro's number Think $\gamma = 1$ i.e. V = 1/N <sub>A</sub>
initial states A   A     A	initial quantities $\#A_0$	initial concentrations $[A]_0$ with $[A]_0 = #A_0/\gamma$	ODE = ODE
A @r A'	A ⊶• A'	$A \rightarrow^k A'$ with $k = r$	Continuous Chemistry Process
A ?a A' B !a B' B'	A+B ⊶• A'+B'	$A+B \rightarrow^k A'+B'$ with $k = r\gamma$	Discrete Chemistry CTMC = CTMC
?a A !a A' @r A"	A+A• <sup>2</sup> r A'+A″	A+A $\rightarrow^{2k}$ A'+A" with k = r $\gamma/2$	Automata are n <sup>2</sup> more compact!
	CTMC	([A]* = d[A]/dt CDE change of concentration over time)	uca Cardelli

# Examples of Chemical Kinetics by Interacting Automata

# Zero-Order Regime

## Or: build me a population like this:



## Second-order and Zero-order Regime



#### Second-Order Regime d[S]/dt = -r[E][S]



E+S	$\rightarrow$	r ES+P
ES –	→s	Е

directive sample 1000.0 directive plot S(); P(); E()

new a@1.0:chan()

let E() = !a; delay@1.0; E() and S() = ?a; P() and P() = ()

run (1 of E() | 1000 of S())

Zero-Order Regime  $d[S]/dt \cong -1$  (by assuming d[ES]/dt=0)



## Cascades





Second-Oder Regime cascade: a signal amplifier (MAPK) aHi > 0 ⇒ cHi = max

directive sample 0.03
directive plot la; lb; lc
new a@1,0:chan new b@1,0:chan new c@1,0:chan
let Amp_hi(a:chan, b:chan) =
do !b; Amp_hi(a,b) or delay@1.0; Amp_lo(a,b)
and Amp_lo(a:chan, b:chan) =
?a; <mark>?a</mark> ; Amp_hi(a,b)
run 1000 of (Amp_lo(a,b)   Amp_lo(b,c))
let A() = !a; A()
run 100 of A()





Zero-Oder Regime cascade: a signal *divider!* aHi = max ⇒ cHi = 1/3 max

directive sample 0.03 directive plot la; lb; lc
new a@1,0:chan new b@1,0:chan new c@1,0:chan
let Amp_hi(a:chan, b:chan) = do lb: delay@1.0: Amp_hi(a,b) or delay@1.0: Amp_lo(a,b) and Amp_lo(a:chan, b:chan) = ?a: ?a: Amp_hi(a,b)
run 1000 of (Amp_lo(a,b)   Amp_lo(b,c))
let A() = la; delay@1.0; A() run 2000 of A()

## Ultrasensitivity



directive sample 215.0	
directive plot S(); P(); E();	ES(): F(): FP()
new a@1.0:chan() new b@1	.0:chan()
let S() = 20: P()	
and P() = ?b; S()	
let E() = la: delav@1.0; E()	
and F() = !b; delay@1.0; F()	)
run 1000 of S()	
let clock(t:float, tick:chan	) = (* sends a tick every t time *)
(val ti = t/100.0 val d = 1.	0/ti (* by 100-step erlang timers *)
let step(n:int) = if n<=0 1	then !tick; clock(t,tick) else delay@d; step(n-
let Sig(pproc() tick:chan)	- (n()   Otick: Sig(n tick))
let raising(p:proc(), rick(char)	= (p()   Flick, Sig(p, lick))
(new tick:chan run (clock)	(t,tick)   Sig(p,tick)))
run 100 of E()	
101100 011()	

#### Zero-Order Regime

A small E-F inbalance causes a much larger S-P switch.



#### Second-Order Regime

No switching behavior



### Or: build me a population like this:



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## Nonlinear Transition (NLT)



## Two NLTs: Bell Shape

9998



## NLT in a Cycle: Oscillator





directive	sample 0.03 1000
directive	plot A(); B(); C()

new a@1.0:chan new b@1.0:chan new c@1.0:chan let A() = do !a;A() or ?b; B() and B() = do !b;B() or ?c; C() and C() = do !c;C() or ?a; A()

run (900 of A() | 500 of B() | 100 of C())





## NLTs in Series: Soliton Propagation





#### directive sample 0.1 1000 directive plot A1(); A2(); A3(); A4(); A5(); A6(); A7(); A8(); A9(); A10(); A11(); A12(); A13() val r=1.0 val s=1.0 new a2@s:chan new a3@s:chan new a4@s:chan new a5@s:chan new a6@s:chan new a7@s:chan new a8@s:chan new a9@s:chan new a10@s:chan new a11@s:chan new a12@s:chan new a13@s:chan let A1() = do delay@r;A2() or ?a2; A2() and A2() = do la2; A2() or delay@r; A3() or ?a3; A3() and A3() = do la3; A3() or delay@r; A4() or ?a4; A4() and A4() = do !a4;A4() or delay@r;A5() or ?a5; A5() and A5() = do !a5; A5() or delay@r; A6() or ?a6; A6() and A6() = do !a6;A6() or delay@r;A7() or ?a7; A7() and A7() = do !a7; A7() or delay@r; A8() or ?a8; A8() and A8() = do !a8;A8() or delay@r;A9() or ?a9; A9() and A9() = do !a9; A9() or delay@r; A10() or ?a10; A10() and A10() = do !a10; A10() or delay@r; A11() or ?a11; A11() and A11() = do !a11; A11() or delay@r; A12() or ?a12; A12() and A12() = do la12;A12() or delay@r;A13() or ?a13; A13() and A13() = !a13;A13()

run 1000 of A1()

# Lotka-Volterra

Or: beyond automata



## **Predator-Prey**



directive sample 1.0 1000 directive plot Carnivor(); Herbivor()

val mortality = 100.0 val breeding = 300.0 val predation = 1.0 new cull @predation:chan()

let Herbivor() =
 do delay@breeding; (Herbivor() | Herbivor())
 or ?cull; ()

and Carnivor() = do delay@mortality; () or !cull; (Carnivor() | Carnivor())

run 100 of Herbivor() run 100 of Carnivor()

An *unbounded state* system!

Plotting: Live

## Lotka-Volterra in Matlab

$H = \tau_{b}; (H H) \oplus ?c_{(p)}; 0$ $C = \tau_{m}; 0 \oplus !c_{(p)}; (C C)$ $\#H_{0}, \#C_{0}$	m=100.0 b=300.0 p=1.0	directive sample 0.35 1000 directive plot Carnivor(); Herbivor() val mortality = 100.0 val breeding = 300.0 val predation = 1.0 new cull @predation:chan()
$ \begin{array}{c} H \rightarrow^{b} H + H \\ C \rightarrow^{m} 0 \\ H + C \rightarrow^{p\gamma} C + C \\ [H]_{0} = \#H_{0}/\gamma \\ [C]_{0} = \#C_{0}/\gamma \end{array} $	γ=1.0 #H <sub>0</sub> = 100 #C <sub>0</sub> = 100	<pre>let Herbivor() =     do delay@breeding; (Herbivor()   Herbivor())     or ?cull; ()     and Carnivor() =     do delay@mortality; ()     or !cull; (Carnivor()   Carnivor())     run 100 of Herbivor()     run 100 of Carnivor()</pre>
$\begin{bmatrix} [H]^{\bullet} = b[H] - p\gamma[H][C] \\ [C]^{\bullet} = -m[C] + p\gamma[H][C] \\ [H]_{0} = \#H_{0}/\gamma \\ [C]_{0} = \#C_{0}/\gamma \end{bmatrix}$	900 900 900 900 900 900 900 900 900 900	
	Extinction	No extinction
	Which one is th	e "right prediction"?

# Biochemistry

## **Or:** Interaction + Complexation

## $\searrow$ $\checkmark$ $\leftrightarrow$ $\checkmark$

Without complexation, many "finite" combinatorial systems can only be expressed by an infinite number of elementary chemical reactions.

## Polyautomata

### **Two new operations** the current states S,T carry an "associaton history"







## Can be encoded in $\pi$ -calculus (and SPiM) by bound-output/bound-input.

directive sample 0.005 directive plot Af(); Ab(); Bf(); Bb()	
val mu = 1.0 val lam = 1.0 new a@mu:chan(chan)	
let Af() = (new n@lam:chan run !a(n); Ab(n)) and Ab(n:chan) = !n; Af()	
let Bf() = ?a(n); Bb(n) and Bb(n:chan) = ?n; Bf()	
run (1000 of Af()   500 of Bf())	

## (Compositional) Enzyme Kinetics



F+S	$ES \xrightarrow{r_2}$	E+P
r <sub>1</sub>		



a@r<sub>0</sub>,r<sub>1</sub>,r<sub>2</sub>

directive sample 0.05 1000 directive plot Ef(); Eb(); Sf(); Sb(); P()
val k1 = 1.0 val km1 = 1.0 val k2 = 100.0 new a@k1:chan(chan,chan)
let P() = ()
let Ef() = (new n@km1:chan new m@k2:chan run !a(n,m); Eb(n,m)) and Eb(n:chan,m:chan) = do !n; Ef() or !m; Ef()
let Sf() = ?a(n,m); Sb(n,m) and Sb(n:chan,m:chan) = do ?n; Sf() or ?m; P()
run (1000 of Ef()   2000 of Sf())



Bidirectional Polymerization



### Actin-like Poly/Depolymerization

new c@µ A<sub>free</sub> =  $!c(v|ft_{\lambda}); A_{b|ft}(|ft)) +$ ?c(rht); A<sub>brht</sub>(rht)  $A_{blft}(lft) =$ !lft; A<sub>free</sub> + ?c(rht); A<sub>bound</sub>(lft,rht) A<sub>brht</sub>(rht) = ?rht; A<sub>free</sub> A<sub>bound</sub>(lft,rht) = !lft; A<sub>brht</sub>(rht)



# Conclusions

# **Compactness of Representation**

- $E_n$  has 2n variables (nodes) and 2n terms (arcs). -  $Ch(E_n)$  has 2n species and n<sup>2</sup> reactions.
- The stoichiometric matrix has size  $2n \cdot n^2 = 2n^3$ .
- The ODEs have 2n variables and  $2n(n+n) = 4n^2$  terms (number of variables times number of accretions plus depletions when sums are distributed)

STOIC	StoichiometricMatrix(Ch(E3))								
	<b>a</b> <sub>00</sub>	<b>a</b> <sub>01</sub>	<b>a</b> <sub>02</sub>	<b>a</b> <sub>10</sub>	a <sub>11</sub>	<b>a</b> <sub>12</sub>	<b>a</b> <sub>20</sub>	<b>a</b> <sub>21</sub>	a <sub>22</sub>
X <sub>0</sub>	-1	-1	-1				+1	+1	+1
$X_1$	+1	+1	+1	-1	-1	-1			
<b>X</b> <sub>2</sub>				+1	+1	+1	-1	-1	-1
<b>Y</b> <sub>0</sub>	-1		+1	-1		+1	-1		+1
<b>Y</b> <sub>1</sub>	+1	-1		+1	-1		+1	-1	
<b>Y</b> <sub>2</sub>		+1	-1		+1	-1		+1	-1

#### $ODE(E_3)$

 $E_3$ 

 $d[X_0]/dt = -r[X_0][Y_0] - r[X_0][Y_1] - r[X_0][Y_2] + r[X_2][Y_0] + r[X_2][Y_1] + r[X_2][Y_2]$  $d[X_1]/dt = -r[X_1][Y_0] - r[X_1][Y_1] - r[X_1][Y_2] + r[X_0][Y_0] + r[X_0][Y_1] + r[X_0][Y_2]$  $d[X_2]/dt = -r[X_2][Y_0] - r[X_2][Y_1] - r[X_2][Y_2] + r[X_1][Y_0] + r[X_1][Y_1] + r[X_1][Y_2]$  $d[Y_0]/dt = -r[X_0][Y_0] - r[X_1][Y_0] - r[X_2][Y_0] + r[X_0][Y_2] + r[X_1][Y_2] + r[X_2][Y_2]$  $d[Y_1]/dt = -r[X_0][Y_1] - r[X_1][Y_1] - r[X_2][Y_1] + r[X_0][Y_0] + r[X_1][Y_0] + r[X_2][Y_0]$  $d[Y_2]/dt = -r[X_0][Y_2] - r[X_1][Y_2] - r[X_2][Y_2] + r[X_0][Y_1] + r[X_1][Y_1] + r[X_2][Y_1]$ 



## Continuous vs. Discrete Kinetics



## Conclusions

- Compositional models
  - Accurate (at the "appropriate" abstraction level).
  - Manageable (so we can scale them up by composition).
  - Executable (stochastic simulation).
- Analysis techniques
  - Mathematical techniques: Markov theory, Chemical Master Equation, and Rate Equation
  - Computing techniques: Abstraction and Refinement, Model Checking, Causality Analysis.
- Many lines of extensions
  - Parametric processes for model factorization
  - Ultimately, rich process-algebra based modeling languages.
- Quantitative techniques
  - Important in the "real sciences".