

Biochemical Systems as Reactive Systems

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Processes and Functions

Functions

$$f(x) = \sqrt{x}$$



“read input into x ; then write \sqrt{x} to output”

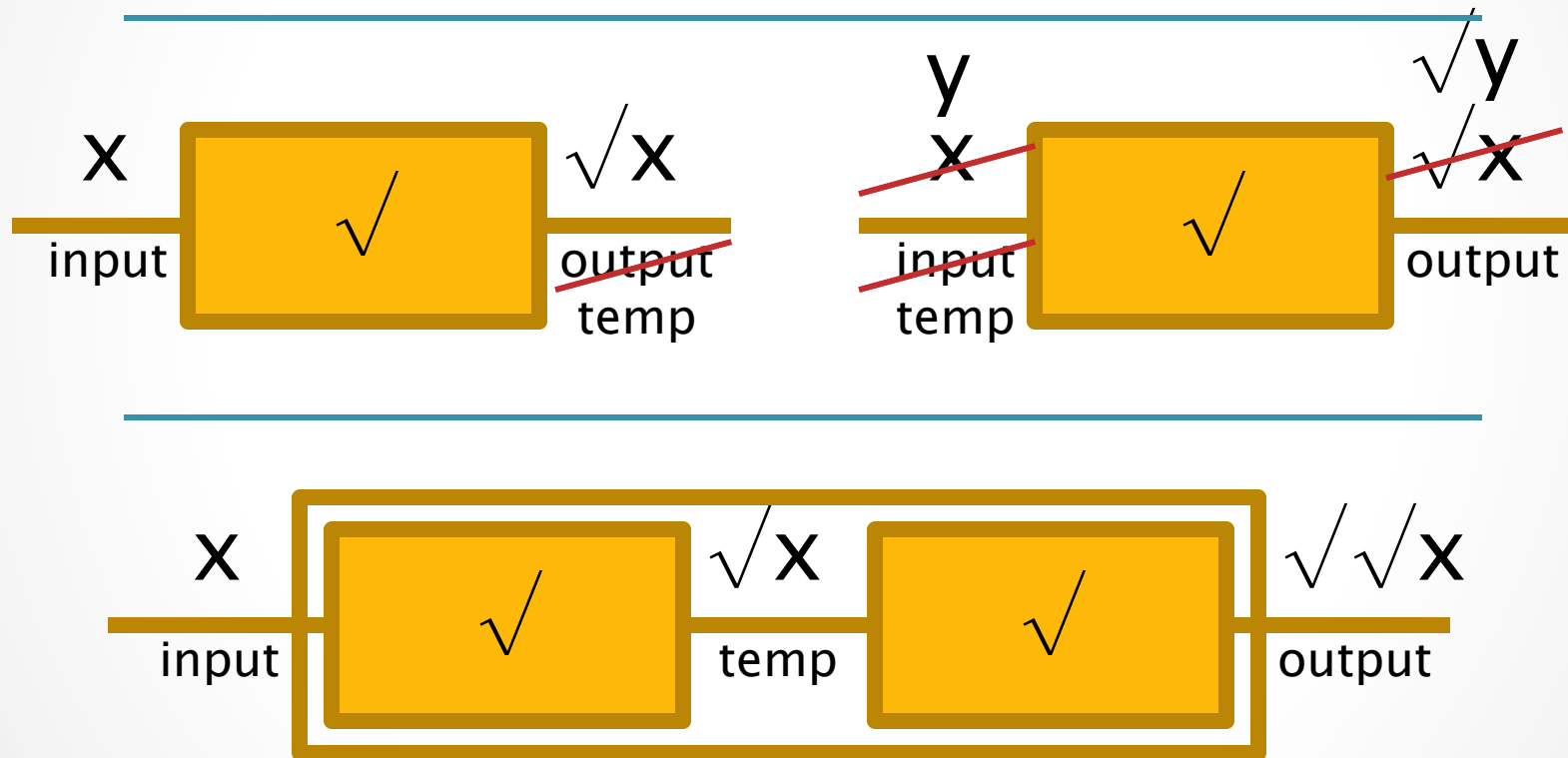
function *read* *write* *channels*

$f = ?input(x); !output(\sqrt{x})$

(binding) input variable *output expression* *(bound) variable occurrence*

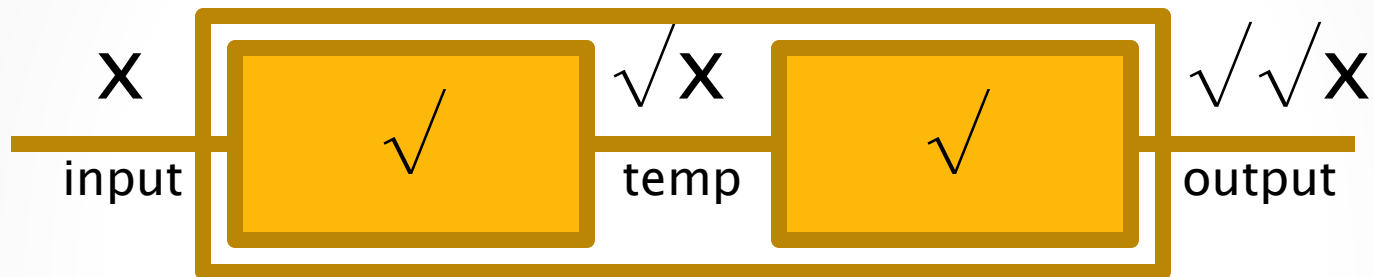
Composing Functions

$$g(x) = (f \circ f)(x) \quad (= f(f(x)))$$



Composing Functions

$$g(x) = (f \circ f)(x)$$



“create a *new* channel and use it to compose two copies of *f*”

channel creation (restriction/hiding/boxing)

(parallel/process) composition

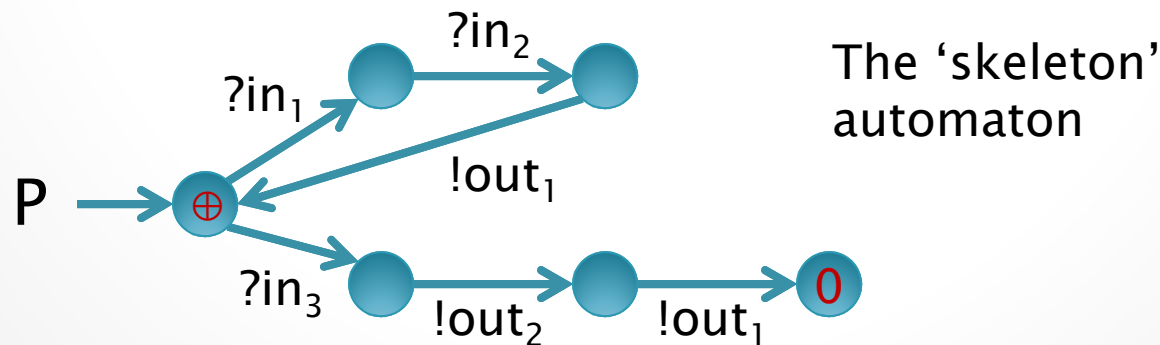
$g = (\nu \text{ temp})$

$?input(x); !temp(\sqrt{x}) \mid$
 $?temp(y); !output(\sqrt{y})$

Many inputs and outputs



$$P = \begin{array}{l} ?in_1(x); ?in_2(y); !out_1(x+y); P \\ \oplus \\ ?in_3(z); !out_2(\sqrt{z}); !out_1(2z); 0 \end{array}$$



That's π -calculus

- To compose processes P we need:
 - Composition: $P \mid P$ (with identity elem. 0)
 - Channel cration: $(\nu x) P$ (with x bound in P)
 - Recursion: $*P$ (equal to $P \mid *P$)
- To execute actions we need:
 - Channel reading: $?c(x); P$ (with x bound in P)
 - Channel writing: $!c(M); P$ (with message M)
 - Choice: $P \oplus P$ (with identity elem. 0)
- ... and channels can be sent as messages!

Generalizing Functions and Automata

- Unlike functions...
 - Processes have multiple, explicitly named, input and output channels.
 - Processes can run in *parallel*, can *deadlock* on their inputs, and can be *nondeterministic* in their outputs.
- Unlike automata (FSA)...
 - Processes can transmit data (not just change state).
 - While automata ‘talk’ to input strings, processes ‘talk’ to other processes: processes are communicating automata.
 - Processes are not “finite state”; they can express unbounded computation in time (divergence) and space (proliferation).

Algebraic Properties

- Functions have one binder and one rule:

- Function application:

If $f(x) =_{\text{def}} M\{x\}$ then $f(a) = M\{a/x\}$

- Processes have two binders and two rules:

- Communication (input '?' binder)

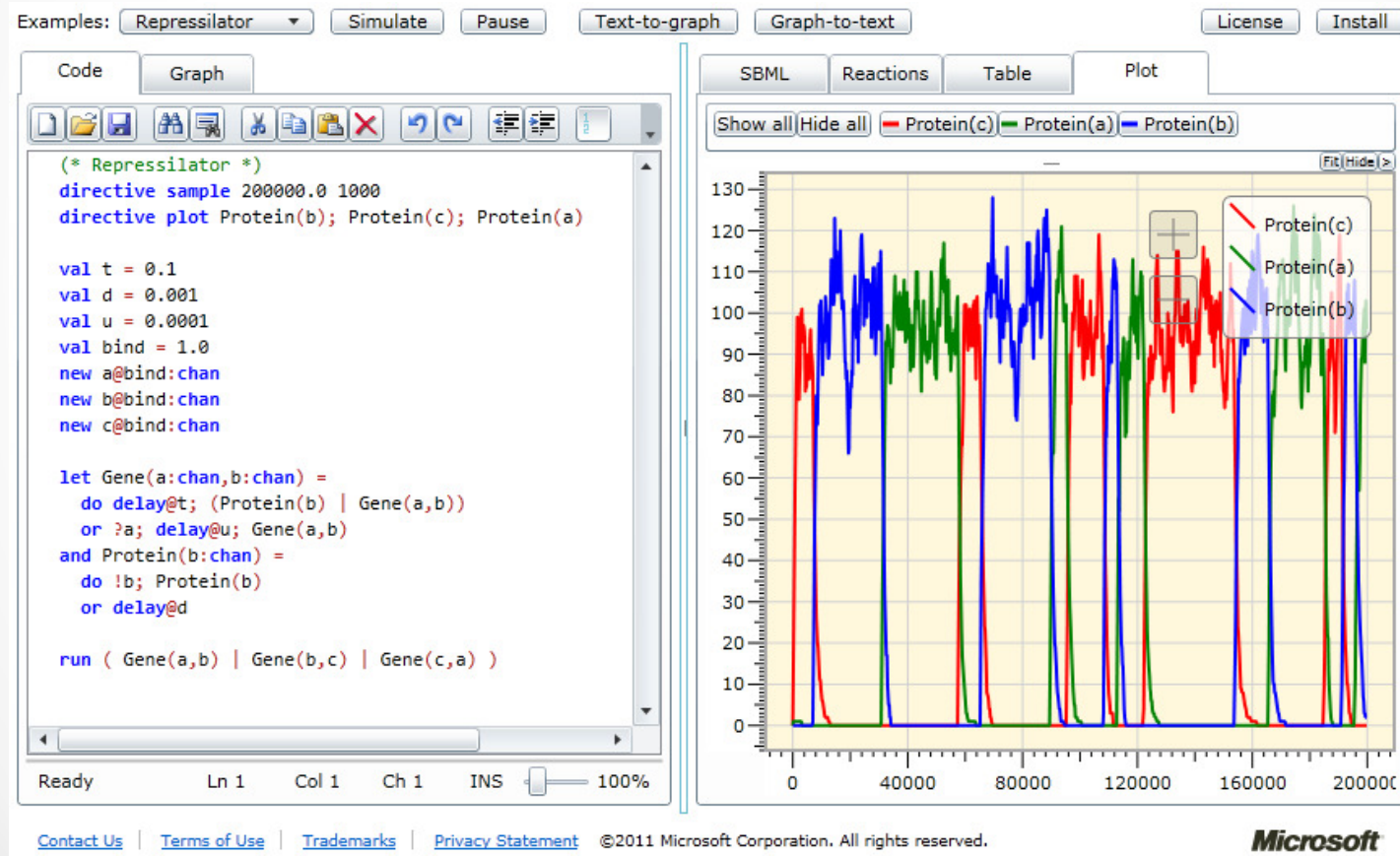
$(?c(x);P\{x\}) \oplus P' \mid (!c(a);Q) \oplus Q' = P\{a/x\} \mid Q$

- Scope extrusion (new 'v' binder)

If x not occurring in Q then $((v x)P) \mid Q = (v x)(P \mid Q)$

Implementations

- SPiM (Stochastic Pi Machine)
 - <http://lepton.research.microsoft.com/VisualSPiM/>
 - Runs in a browser with Silverlight.



Processes and Chemistry

Continuous Chemical Systems

Reactions:



Degradation



Asymmetric Collision



Symmetric Collision

Continuous reaction kinetics, respectively:

$$[A]^\bullet = -r[A]$$

Exponential Decay

$$[A_i]^\bullet = -r[A_1][A_2]$$

Mass Action Law

$$[A]^\bullet = -2r[A]^2$$

Mass Action Law

(assuming $A \neq B_i \neq A_j$ for all i, j)

π -calculus for Chemistry

- To compose *soups* P we need:

- Stochastic channels: $(\nu x_r) P$ r is the rate of an exponential distribution: the rate of communication on that channel
- Composition: $P \mid P$ (with identity elem. 0)
- Recursion: $*P$ (equal to $P \mid *P$)

- To execute *species* we need:

- Collision: $?x_r; P$ (with no input variables)
- Co-collision: $!x_r; P$ (with no output messages)
- Delay: $\tau_r; P$ ($= (\nu x_r) ?x_r; P \mid !x_r; 0$ for any x not in P)
- Choice: $P \oplus P$ (with identity elem. 0)

Discrete Chemical Systems (1)

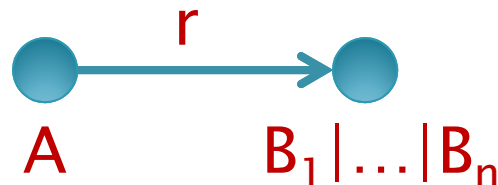
Reaction:



Discrete reaction kinetics:

$$A = \tau_r; (B_1 | \dots | B_n)$$

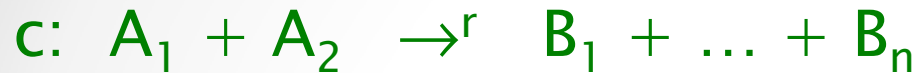
The mathematical meaning of that is a Continuous Time Markov Chain (for a specific set of initial conditions, e.g. a single A molecule), here represented as a transition graph:



Hence the π -calculus description abstracts from initial conditions (like ODEs). For each set of initial conditions, a CTMC can be systematically extracted from the stochastic π -calculus models.

Discrete Chemical Systems (2)

(Uniquely named) reaction:

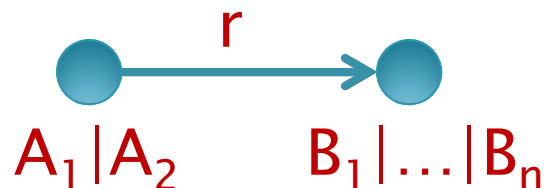


Discrete reaction kinetics:

$A_1 = ?c_r; (B_1 | \dots | B_i)$ (the name of the reaction becomes the channel)

$A_2 = !c_r; (B_i | \dots | B_n)$ (splitting results is arbitrary: $1 \leq i \leq n$)

With initial conditions $A_1 | A_2$ (single molecules), the CTMC is:



Discrete Chemical Systems (3)

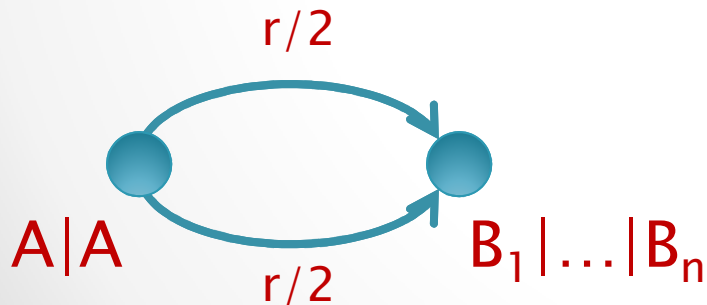
(Uniquely named) reaction:



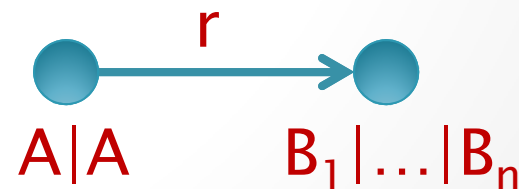
Discrete reaction kinetics:

$$A = ?c_{r/2}; (B_1 | \dots | B_i) \oplus !c_{r/2}; (B_i | \dots | B_n) \quad 1 \leq i \leq n$$

With initial conditions $A|A$ (two molecules), the CTMC is as follows; note that each copy of A can do an input or an output, so there are two possible paths to the outcome:



That is:



From Reactions to Processes



Interaction Matrix

channels
(1 per reaction)

Half-rate for symmetric reactions

processes
(1 per species)

	$v_{1(k1)}$	$v_{2(k2)}$	$v_{3(k3)}$	$v_{4(k4/2)}$
A	$?:(C C)$	$?;D$		
B	$!;0$			
C		$!;0$	$\tau:(E F)$	
D				
E				
F				$?;B$ $!;0$

Fill the matrix by columns:

Degradation reaction $v_i: X \xrightarrow{k_i} P_i$
add $\tau;P_i$ to $\langle X, v_i \rangle$.

Asymmetric reaction $v_i: X+Y \xrightarrow{k_i} P_i$
add $?;P_i$ to $\langle X, v_i \rangle$ and $!;0$ to $\langle Y, v_i \rangle$

Symmetric reaction $v_i: X+X \xrightarrow{k_i} P_i$
add $?;P_i$ and $!;0$ to $\langle X, v_i \rangle$

Read out the processes by rows:

$$A = ?v_{1(k1)}:(C|C) \oplus ?v_{2(k2)};D$$

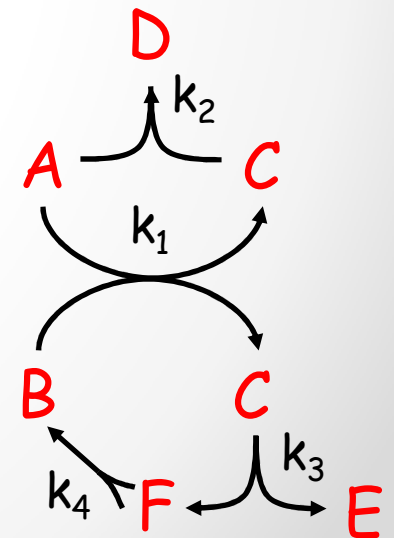
$$B = !v_{1(k1)};0$$

$$C = !v_{2(k2)};0 \oplus \tau_{k3};(E|F)$$

$$D = 0$$

$$E = 0$$

$$F = ?v_{4(k4/2)};B \oplus !v_{4(k4/2)};0$$



That Chemical System in SPiM

$A = ?v_{1(k1)}; (C|C) \oplus ?v_{2(k2)}; D$

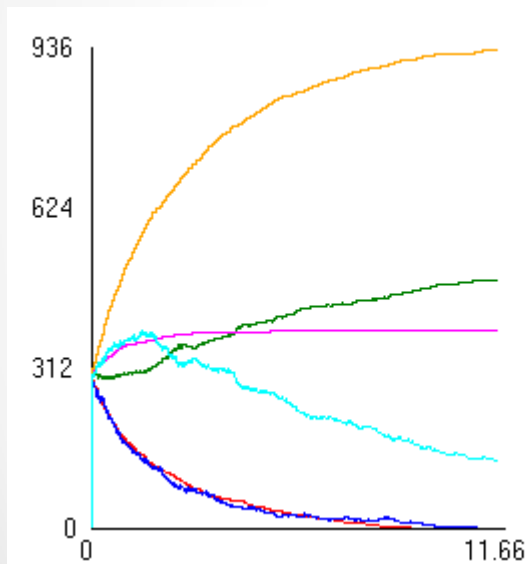
$B = !v_{1(k1)}; 0$

$C = !v_{2(k2)}; 0 \oplus \tau_{k3}; (E|F)$

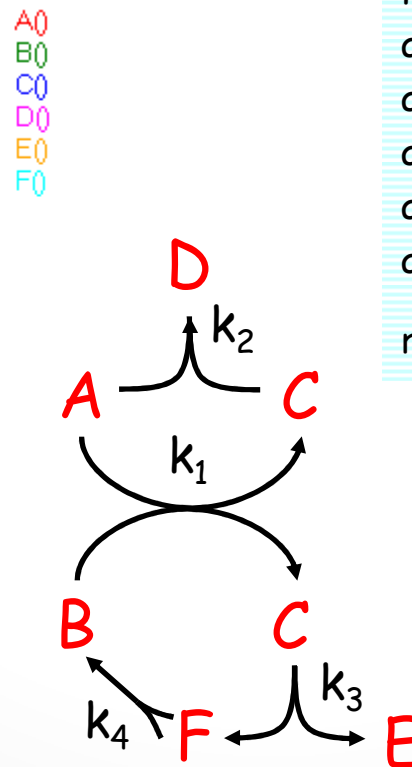
$D = 0$

$E = 0$

$F = ?v_{4(k4/2)}; B \oplus !v_{4(k4/2)}; 0$



Gillespie-style
stochastic simulation



```
directive sample 10.0
```

```
directive plot A(); B(); C(); D(); E(); F()
```

```
val k1 = 0.001 new v1@k1:chan
```

```
val k2 = 0.001 new v2@k1:chan
```

```
val k3 = 1.0
```

```
val k4 = 0.001 new v4@k4/2.0:chan
```

```
let A() = do ?v1;(C()|C()) or ?v2;D()
```

```
and B() = !v1
```

```
and C() = do !v2 or delay@k3;(E()|F())
```

```
and D() = ()
```

```
and E() = ()
```

```
and F() = do ?v4;B() or !v4
```

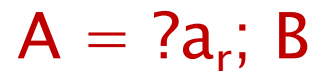
```
run 300 of (A()|B()|C()|D()|E()|F())
```

Modeling Techniques

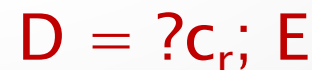
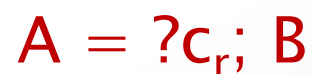
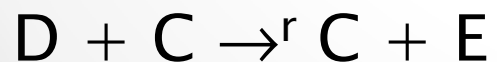
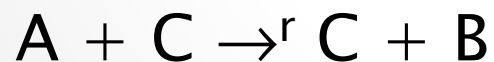
- That is a *systematic* way to translate reactions to processes.
- But there can be *better* ways to do it.
- That is, ways that produce *more compact and/or modular models*, but with the *same kinetics*.

Ex: Catalysis

- Two reactions, same catalyst C
 - According to the general scheme the catalyst uses one channel for each reaction it catalyzes

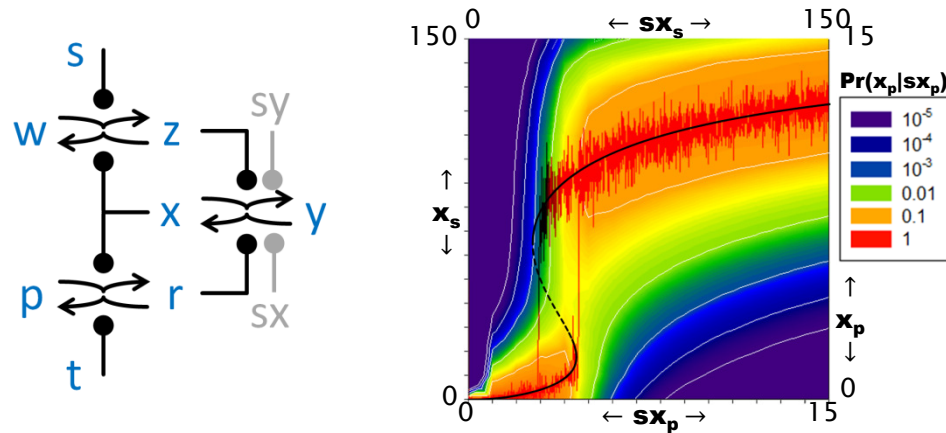


- Modularizing: the catalyst has its own catalysis channel c , used for all the reactions it catalyzes:



Ex: System Analysis

- 16-reaction abstract model of cell-cycle switch.



sx (hor axis): input value
 x (ver axis): output value at equilibrium

Black line: deterministic bifurcation diagram

Red line: stochastic simulation (sx_s, x_s) at size $x_{\max} = 150$ [by SPiM]

Heatmap: discrete probability distribution (sx_p, x_p) at size $x_{\max} = 15$ [by PRISM]

(Joint work with Attila Csikasz-Nagy)

Processes and Biochemistry

π -calculus for Biochemistry

- Biochemistry here means: direct modeling of complexation and polymerization.
- We now go back to the full (and stochastic) π -calculus: we need to pass channels as messages!

Complexation



There is no good notation for this reaction in chemistry: $A:B$ is considered as a separate species (which leads to combinatorial explosion of models).

But there is a way to write this precisely in π -calculus. There is a single public *association* channel a_r at rate r , and many private *dissociations* channels d_s at rate s , one for each complexation event (created by ν):

$$\begin{aligned} A_{\text{free}} &= (\nu d_s) !a_r(d_s); A_{\text{bound}}(d_s) \\ A_{\text{bound}}(d_s) &= !d_s; A_{\text{free}} \end{aligned}$$

$$\begin{aligned} B_{\text{free}} &= ?a_r(d_s); B_{\text{bound}}(d_s) \\ B_{\text{bound}}(d_s) &= ?d_s; B_{\text{free}} \end{aligned}$$

Note that we are describing A *independently* of B : as in the catalysis example, A could form complexes with many different species over the a_r channel.

More compactly:

$$\begin{aligned} A &= (\nu d_s) !a_r(d_s); !d_s; A \\ B &= ?a_r(d_s); ?d_s; B \end{aligned}$$

Polymerization

- Polymerization is iterated complexation
 - It can be represented in π -calculus *finitely*, with **one process (definition) for each monomer**.
 - Note that polymerization cannot be described *finitely* in chemistry (or ODEs) because there it needs one reaction for each *length* of polymer.
 - The reason it works in π -calculus is because of the ν operator. It enables the finite representation of systems of potentially unbounded complexity.
 - Like real biochemistry, where the structure of each monomer is coded in a finite piece of DNA, and yet unbounded-length polymers happen.

Conclusions

Conclusions

- π -calculus

- A mathematical notation for reactive systems
- In stochastic form, suitable for representing discrete chemistry, biochemistry, etc.
- Some unique properties: ability to finitely express systems of unbounded complexity, like networks of complexing proteins.

- Further Reading

- R. Milner: **Communicating and Mobile Systems: The Pi Calculus**
- A. Regev, E. Shapiro. **Cellular Abstractions: Cells as Computation**. NATURE vol 419, 2002-09-26, 343.
- L. Cardelli: **From Processes to ODEs by Chemistry**. TCS 2008, DOI: http://dx.doi.org/10.1007/978-0-387-09680-3_18
- A. Phillips, L. Cardelli, **A Correct Abstract Machine for the Stochastic Pi-calculus**, in *Concurrent Models in Molecular Biology*, 2004.